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Key indicators

Single-crystal X-ray study
 $T = 294$ K
Mean $\sigma(\text{C}-\text{C}) = 0.004$ Å
 R factor = 0.069
 wR factor = 0.159
Data-to-parameter ratio = 12.6For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.**(E)-1-(4-Benzoyloxy-3-methoxybenzylidene)-
2-(4-nitrophenyl)hydrazine**

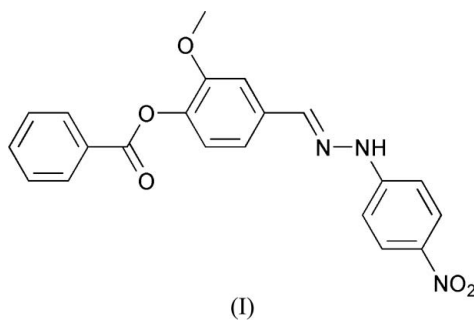
In the title compound, $\text{C}_{21}\text{H}_{17}\text{N}_3\text{O}_5$, the central planar group makes dihedral angles of 83.55 (11) and 5.54 (8) $^\circ$ with the terminal phenyl ring and the phenylhydrazine mean planes, respectively. The molecular packing is stabilized by intermolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds that link adjacent molecules into one-dimensional chains.

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Comment

A large number of Schiff base derivatives have been synthesized and employed in the development of protein and enzyme mimics, for example as a model to mimic hydrolase in the hydrolysis of *p*-nitrophenyl picolinate (Li *et al.*, 2005).

In the title compound, (I) (Fig. 1), a nitrophenylhydrazine Schiff base derivative, the central group (C8–C13/C15/O2/O3) is essentially planar, with an r.m.s. deviation of the fitted atoms of 0.01 Å. It makes dihedral angles of 5.54 (8) and 83.55 (11) $^\circ$ with the phenylhydrazine residue and with the terminal phenyl ring, respectively. The nitro group is twisted around the C–N bond; the dihedral angle between the nitro group and the C16–C21 ring is 15.73 (16) $^\circ$.



The crystal packing of (I) is stabilized by intermolecular $\text{N}2-\text{H}2\cdots\text{O}5^i$ [symmetry code (i): $\frac{5}{2} - x, \frac{1}{2} + y, \frac{5}{2} - z$] hydrogen

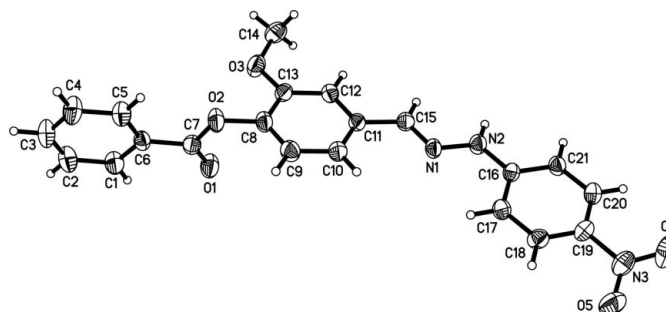


Figure 1
The molecular structure of (I), with displacement ellipsoids for non-H atoms drawn at the 30% probability level.

bonds that link adjacent molecules into one-dimensional chains running along the *b* axis (Table 1 and Fig. 2).

Experimental

An anhydrous ethanol solution (50 ml) of 4-formyl-2-methoxyphenyl benzoate (2.56 g, 10 mmol) was added to an anhydrous ethanol solution (50 ml) of 1-(4-nitrophenyl)hydrazine (1.53 g, 10 mmol) and the mixture stirred at 350 K for 5 h under nitrogen, giving an orange precipitate. The product was isolated, recrystallized from acetonitrile and then dried in a vacuum to give the pure compound in 78% yield. Orange single crystals of (I) suitable for X-ray analysis were obtained by slow evaporation of an acetonitrile solution.

Crystal data

$C_{21}H_{17}N_3O_5$	$Z = 4$
$M_r = 391.38$	$D_x = 1.361 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation
$a = 8.1110 (16) \text{ \AA}$	$\mu = 0.10 \text{ mm}^{-1}$
$b = 14.965 (3) \text{ \AA}$	$T = 294 (2) \text{ K}$
$c = 16.220 (3) \text{ \AA}$	Block, orange
$\beta = 104.07 (3)^\circ$	$0.15 \times 0.12 \times 0.10 \text{ mm}$
$V = 1909.7 (7) \text{ \AA}^3$	

Data collection

Bruker SMART APEX CCD area-detector diffractometer	11378 measured reflections
φ and ω scans	3322 independent reflections
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	1960 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.973$, $T_{\max} = 0.990$	$R_{\text{int}} = 0.082$
	$\theta_{\text{max}} = 25.0^\circ$

Refinement

Refinement on F^2	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.069$	$w = 1/[\sigma^2(F_o^2) + (0.0709P)^2]$
$wR(F^2) = 0.159$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.02$	$(\Delta/\sigma)_{\text{max}} < 0.001$
3322 reflections	$\Delta\rho_{\text{max}} = 0.15 \text{ e \AA}^{-3}$
263 parameters	$\Delta\rho_{\text{min}} = -0.17 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$N2-H2\cdots O5^i$	0.86	2.06	2.907 (4)	169

Symmetry code: (i) $-x + \frac{5}{2}, y + \frac{1}{2}, -z + \frac{5}{2}$.

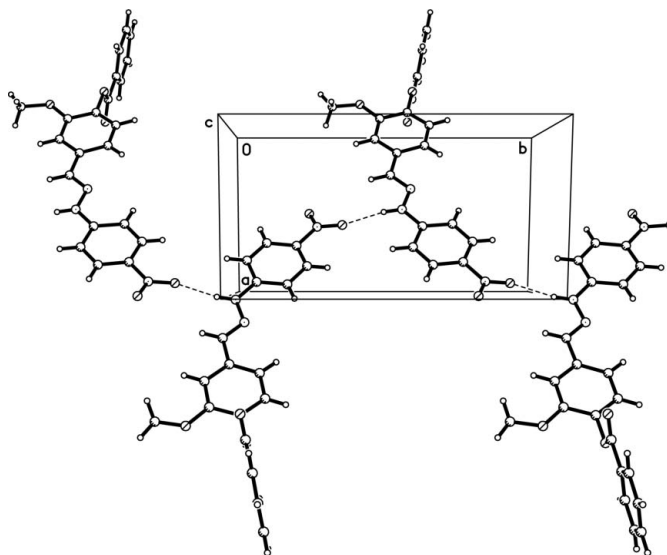


Figure 2

Part of the crystal structure of compound (I), showing the formation of a hydrogen-bonded chain. Hydrogen bonds are drawn as dashed lines.

The H atoms were included in calculated positions and refined using a riding-model approximation. Constrained C–H and N–H bond lengths and isotropic $U_{\text{iso}}(\text{H})$ parameters: 0.93 \AA and $1.2U_{\text{eq}}(\text{C})$ for C_{sp^2} , 0.96 \AA and $1.5U_{\text{eq}}(\text{C})$ for methyl, and 0.86 \AA and $1.2U_{\text{eq}}(\text{N})$ for imino H atoms.

Data collection: SMART (Bruker, 1999); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997a); molecular graphics: SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: SHELXTL.

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